

Rapiflex



Solid Sample Preparation – **Evaporative Grinding**



Solution Sample Preparation:

- DCTB at 60 mg/ml, NaTFA at 5 mg/ml and Polymer at 10 mg/ml
- Mix in a 20:1:1 ratio (Matrix:Cation:Polymer)
- Spot 0.5 μl onto the target with gentle blowing

Solid Sample Preparation – Evaporative Grinding:

- Place a small amount of sample on target, add a very small amount NaTFA, add a slight excess of matrix
- Add 10 μ l of acetone and grind on the target surface, scrape of the excess to a thin layer and remove the powder with clean air

Data Processing:

Data analysis was accomplished using Polytools (Bruker) and Polymerix from Sierra Analytics

Mechanopolymers and Vitrimers:

- Mechanopolymers are unique materials that possess mechanically labile bonds allowing them to change their properties upon the application of external stimuli (degradable on demand)
 - light, heat, pH, or redox potential
- This ability has led to various useful properties such as color/fluorescence change, isomerization and efficient degradation pathways

- Vitrimers are compounds derived from thermosets and are very similar to them
- Vitrimers consist of molecular, covalent networks that change their topology by thermally activated bondexchange reactions
- At high temperatures they can flow like viscoelastic liquids but at low temperatures the bond-exchange reactions are slow, they behave like classical thermosets
- Their behavior opens new possibilities in the application of thermosets like self-healing material or simple processibility in a wide temperature range.





• MALDI MS easily characterizes polymeric/oligomeric samples allowing for complete characterization of synthesized or purchased materials

- contamination of the final material with products resulting from side reactions

MALDI-TOF MS Analysis of Mechanopolymers and Vitrimers

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Vitrimer Component

EPON 828 - Bisphenol A diglycidyl ether:

Complete characterization is necessary to understand synthetic routes of novel materials such as the Mechanopolymer, which helps decrease undesired products Understanding materials used in the synthesis of additional products (vitrimer) is paramount in avoiding impure materials not suitable for reaction and avoiding

The complete characterization of a network like the vitrimer is extremely difficult due to insolubility and desorption/ionization of a large branched molecule Evaporative grinding (solid sample prep) enabled the acquisition of preliminary data showing side and/or degradation products – more work needed to optimize

pyrrolidinedione) readily observed











Fragments:

- The structure shown is the best fit to the observed MS/MS data
- Possibly cyclic?